

**Gavazov K.**, Simeonova Z., Alexandrov A. *Extraction-spectrophotometric study of the system vanadium(V) – 4-(2-pyridylazo)resorcinol – 2,2',5,5'-tetraphenyl-3,3'-(p-biphenyl)ditetrazolium chloride – water – chloroform. Determination of vanadium in steels, **Analytical Laboratory**, 7 (3) (1998) 127-133.*

A ternary ion-association complex of vanadium(V) with 4-(2-pyridylazo)resorcinol (PAR) and 2,2',5,5'-tetraphenyl-3,3'-(p-biphenyl)ditetrazolium chloride (NTC)  $V(V) : PAR : NTC = 2 : 4 : 3$  was obtained, which is extracted into  $CHCl_3$ . The molar absorptivity of the complex at  $\lambda_{max} = 560$  nm was calculated to be  $\epsilon = 3.69 \times 10^4$  l.mol.cm. The optimum conditions for extraction (pH, concentration of reagents, time), as well as the distribution constant ( $\log K_D = 1.26 \pm 0.06$ ), the association constant ( $\log \beta = 7.19 \pm 0.15$ ), the extraction constant ( $\log K_{ex} = 8.45 \pm 0.21$ ), and the recovery factor  $R$  (%) =  $94.8 \pm 0.7$  were determined. The effect of foreign ions in the absence and in the presence of 1,2-diaminocyclohexane-N,N,N',N'-tetraacetic acid (CDTA) was studied. A direct sensitive and selective extraction-spectrophotometric method for determination of vanadium in steels was developed. In the presence of CDTA the molar absorptivity decreases by about 20%, but the determination of vanadium is not interfered by a 10000-fold excess of  $NH_4^+$ ,  $K^+$ ,  $Na^+$ ,  $SO_4^{2-}$ ,  $F^-$ ,  $HPO_4^{2-}$ ,  $H_2PO_4^{2-}$ , tartrate<sup>2-</sup>; 5000-fold excess of citrate<sup>2-</sup>; 2000-fold excess of Cd(II); 1500-fold excess of Pb(II),  $Cl^-$ ; 1000-fold excess of Cr(III), Ni(II), Cu(II), Zn(II), Co(II), Re(VII),  $C_2O_4^{2-}$ ; 800-fold excess of Ce(III); 750-fold excess of Ti; 600-fold excess of Fe(III); 300-fold excess of Al(III); 100-fold excess of Mn(II), Mo(II),  $NO_3^-$ ; 70-fold excess of  $CrO_4^{2-}$ ; 50-fold excess of  $SO_3^{2-}$ ; 5-fold excess of Nb(V); 4-fold excess of V(IV); 2-fold excess of W(VI); W(VI) and Nb(V) can easily be separated during the treatment of the sample. Beer's law is obeyed in the range of 1.5 - 35  $\mu$ g of vanadium in 10 ml of chloroform extract.

**Keywords:** Vanadium, 4-(2-Pyridylazo)resorcinol, 2,2': 5,5'-Tetraphenyl-3,3'-(p-biphenyl)ditetrazolium chloride, Extraction-spectrophotometric determination, Steel